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ELECTRONIC HALL EFFECT IN NaCl

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By applying the pulsed light method of Pohl and his co-workers<sup>1</sup> to the measurement of electronic Hall effect in insulating photoconductors it has been possible to avoid the space charge difficulties present in previous work<sup>2,3</sup>. The electrode arrangement is shown in Figure 1. The battery  $V_L$  sets up a potential gradient on the lower surface of the crystal, and thus an inhomogeneous field in the crystal which is, on the average, in the x-direction. The upper plate is held at ground potential by the electrometer circuit; the potential of the entire lower plate can be changed with the potentiometer, and moving the potentiometer tap rotates the electric field in the crystal. The electrometer responds only to the component of charge flow which is in the y direction.

In making a measurement, a magnetic field is applied in the z direction, the crystal is illuminated with pulses of light a few seconds in duration, and the potentiometer is adjusted until the electrometer gives zero deflection during illumination. This means that the average charge flow in the crystal is in the x direction parallel to the electrodes, but the average electric field is at an angle, the Hall angle, to the electrodes. To eliminate space charge, the batteries are turned off between each observation of the electrometer and the crystal is illuminated with a strong light.

The magnetic field is then reversed and the potentiometer is again adjusted as before, so that the average charge flow is again in the x direction but the average electric field has been rotated through twice the Hall angle. Using a theorem of Shockley<sup>4</sup> it can then be shown rigorously that (for small  $\theta$ )

$$\frac{2\mu H}{c} = 2\theta = \frac{\Delta V}{D} \frac{\int n\tau dv}{\int E_x n\tau dv}$$

where  $\mu$  is the mobility,  $H$  the magnetic field,  $c$  the velocity of light,  $\theta$  the Hall angle,  $D$  the thickness of the crystal,  $n$  the number of electrons per unit volume released by the light,  $\tau$  the mean electronic lifetime, and  $E_x$  the x component of electric field in the crystal. The integrations are over the volume of the crystal.

This method differs from the conventional one in several ways. First, it is a transient method. Second, the electric field inside the crystal is rotated by external electrodes, rather than by the electrons under study. Finally, no attempt is made to inject or eject electrons through the surface of the crystal. The method cannot be applied directly to crystals having finite dark conductivity. It has already been applied to diamond.<sup>5</sup>

Both additively colored and x-rayed Harshaw NaCl crystals have been studied and the charge released from F centers in these crystals gives a Hall voltage polarity characteristic of electrons. At  $82 \pm 2^\circ\text{K}$  the mobility is  $250 \pm 50 \text{ cm}^2/\text{volt sec.}$ , and at  $200^\circ\text{K}$  the mobility is of the order of forty. The experimental error is due to noise in the conducting plates, F center bleaching, temperature variations, or possibly crystal imperfections. By violently quenching the crystal its mobility can be reduced to as low as  $150 \text{ cm}^2/\text{volt sec.}$  at  $82^\circ\text{K}$ . These observations are in reasonable agreement with both the theory of Frohlich and Mott<sup>6</sup> and the more recent calculation of Low and Pines<sup>7</sup>. Work is under way to improve and extend these measurements.

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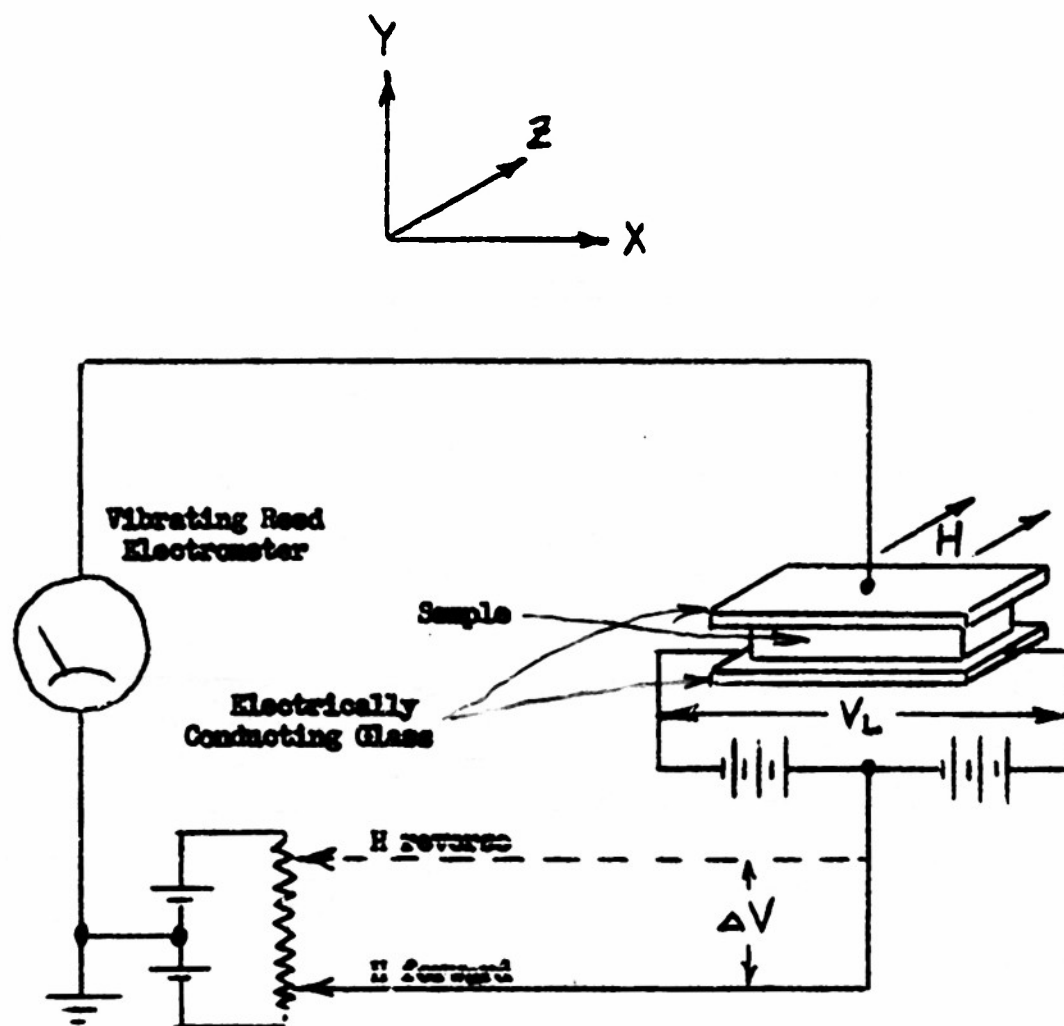


Figure 1. Electrode arrangement for observing electronic Hall effect in insulating photoconductors.